

ORIGINAL RESEARCH ARTICLE

Chemical composition and antibacterial activity of floral volatile oil of *Alstonia scholaris* (L.) R. Br. from India

MUNMUN K SINGH¹ • SWATI SINGH¹ • SANYOGITA YADAV² • PRAVEEN K KASHYAP¹ • SUDEEP TANDON¹ • MAHENDRA P DAROKAR² • RAM S VERMA^{1*}

Key Words

Apocynaceae
Flower volatiles
GC-MS
Linalool
Linalool oxides
Saptaparna

ABSTRACT

Alstonia scholaris (L.) R. Br, commonly known as saptaparna is an evergreen tropical tree of medicinal value. In this study, the floral volatile oil composition and antibacterial activity of *A. scholaris* collected in the morning and evening times were investigated. A total of thirty-six constituents, forming 81.1-83.7 % of total composition were identified. The oil was characterised by the presence of a higher amount of oxygenated monoterpenes (60.3-66.7 %), followed by benzenoids (9.2-9.9 %). Major components of the oil were linalool (13.6-14.3%), cis-linalool oxide (furanoid) (7.3-12.2 %), trans-linalool oxide (furanoid) (7.0-7.5 %), cis-linalool oxide (pyranoid) (3.3-7.2 %), trans-linalool oxide (pyranoid) (3.8-4.7 %), terpinen-4-ol (7.9-10.6 %), and α -terpineol (6.3-6.6 %). The oil showed moderate to good activity against *Salmonella typhimurium* (MTCC 98), while it showed moderate activity against *Escherichia coli* (MTCC 739). It can be concluded that the collection time influenced the yield and chemical composition of *A. scholaris* floral volatiles.

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INTRODUCTION

Alstonia scholaris (L.) R. Br (Apocynaceae), popularly known as 'saptaparna' is an evergreen tropical tree native to the Indian sub-continent and Southeast Asia. The plant is distributed throughout moist regions of India, especially in West Bengal and west-coast forests of southern India. In the Indian system of medicine (Ayurvedic, Unani, and Siddha), the bark of the plant is used as a stimulant, febrifuge, spasmolytic, antidysenteric, uterine, antiperiodic, hypotensive, and for internal fevers

(Khare, 2007). It is used to cure asthma, pneumonia, tuberculosis, cholera, diarrhea, cancer, hepatitis, and malaria (Jain, 1991; Mollik et al., 2010; Kaushik et al., 2011). Phytochemical studies of the leaf, root, and bark of *A. scholaris* have resulted in the isolation of various bioactive molecules of diverse chemical classes like alkaloids, terpenoids, phlobatannins, steroids, and saponins that exhibited various pharmacological activities (Khyade et al., 2014). Several indole alkaloids, including echitamine, echitamidine,

*Corresponding author; Email: rs.verma@cimap.res.in

¹Phytochemistry Division, ²Bioprospection and Product Development Division, CSIR-Central Institute of Medicinal and Aromatic Plants (CIMAP), PO CIMAP, Lucknow 226015, India

Doi: <https://doi.org/10.62029/jmaps.v42i4.Singh>

akuammiginone, scholaricine, 19-*epi*-scholaricine, vallesamine, picrinine have been isolated and characterised from the plant (Keawpradub et al., 1997; Yamauchi et al., 1990a; 1990b; Salim et al., 2004; Zhao et al., 2020). Echitamine is reported as a lipase inhibitor, antiproliferative agent, and targeted for cancer chemotherapy (Saraswathi et al., 1999; Jahan et al., 2009; Singh et al., 2015; George et al., 2019). Besides, *A. scholaris* possesses a pleasant-smelling fragrance in its beautiful flowers. The floral volatile oil composition of the plant has been studied in the past from Vietnam (Dung et al., 2001). However, studies dealing with the chemical composition of the floral volatile oil of *A. scholaris* with reference to collection time are meagre. Therefore, it was planned to examine the floral volatile oil composition and antibacterial activity of *A. scholaris* collected early in the morning and evening time from India.

MATERIALS AND METHODS

Plant material and isolation of floral volatile oil

Fresh flowers of *A. scholaris* were collected in full bloom in the morning (6 am) and evening (6 pm) from Lucknow, Uttar Pradesh in October 2019 (voucher specimen no: CIMPANT1005). The freshly collected flowers (1.0 kg) were immediately subjected to hydrodistillation in duplicates in a Clevenger's type apparatus and floral distillates (1 liter) were collected. The floral distillate was saturated with sodium chloride and then extracted successively with hexane (200 ml) and dichloromethane (200 ml). The organic layers were mixed, dehydrated over anhydrous sodium sulphate, and concentrated under vacuum at 35 °C. The concentrated floral volatiles were collected and stored in the refrigerator (3°C) until further analysis.

GC-FID and GC-MS analysis

The floral volatile oil analysis was carried out using gas chromatography (GC-FID) and gas chromatography-mass spectrometry (GC-MS) techniques as reported earlier (Nogueira et al., 2001; Adams, 2007; Andriamaharavo, 2014; Verma et al., 2020).

Antibacterial assay

Antibacterial activity of floral volatile oils of *A. scholaris* was performed by disc diffusion method against Gram-positive bacteria [*Staphylococcus aureus* (MTCC-96), *Streptococcus epidermidis* (MTCC-435), *Streptococcus mutans* (MTCC-890)], and Gram-negative bacteria [*Salmonella typhimurium* (MTCC-98), *Klebsiella pneumoniae* (MTCC-109), *Escherichia coli* (DH5á), *Escherichia coli* (EC739)] as per CLSI guidelines (Verma et al., 2016).

RESULTS AND DISCUSSION

The yield (w/w) of the floral volatile oil of *A. scholaris* collected in the morning and evening time was 0.01 % and 0.02 %, respectively. Previously, 0.03% yield of volatile concentrate was observed from the flowers of *A. scholaris* (Dung et al., 2001). These variations are possible due to the location, variation of extraction methods adopted, etc (Barra, 2009). The resulting floral volatile oils were analysed by GC-FID and GC-MS. A total of thirty-six constituents, forming 81.1-83.7 % of total composition were identified (Table 1). The oils were characterised by higher amount of oxygenated monoterpenes (66.7 % at 6 am and 60.3 % at 6 pm), followed by benzenoids (9.2% at 6 am and 9.9 % at 6 pm). Major components of the oils were linalool (13.6 % at 6 am and 14.3 % at 6 pm), *cis*-linalool oxide (furanoid) (12.2 % at 6 am and 7.3 % at 6 pm), *trans*-linalool oxide (furanoid) (7.0 % at 6 am and 7.5 % at 6 pm), *cis*-linalool oxide (pyranoid) (7.2 % at 6 am and 3.3 % at 6 pm), *trans*-linalool oxide (pyranoid) (4.7 % at 6 am and 3.8 % at 6 pm), terpinen-4-ol (7.9 % at 6 am and 10.6 % at 6 pm), α -terpineol (6.3 % at 6 am and 6.6 % at 6 pm), and phenyl ethyl alcohol (4.5 % at 6 am and 6.5 % at 6 pm). The structures of the major floral volatiles and a representative chromatogram of floral volatile oil of *A. scholaris* are shown in Figure 1 and Figure 2, respectively. The floral volatile oils extracted from the morning and evening collected flowers showed the presence of similar components, but in different relative compositions. The amount of *cis*-linalool oxides (furanoid and pyranoid), *trans*-linalool oxide (pyranoid), and *p*-vinyl guaiacol were higher in the morning than

evening. However, linalool, *trans*-linalool oxide (furanoid), phenyl ethyl alcohol, terpinen-4-ol, α -terpineol, and hexahydrofarnesyl acetone were relatively higher in the evening than morning. Diurnal variability studies of aromatic plants

showed that the volatile oil composition changes during the day hours. The content of the essential oil is observed to be varying during different hours of day time. Generally, it increases from the morning to afternoon and afterward decreases in

Table 1: Chemical constituents of the floral volatile oil of *Alstonia scholaris*

Compound	RI ^a	RI ^b	Content (%) ^c	
			6.00 am	6.00 pm
Dehydro-1,8-cineole	990	988	3.1	1.8
1,8-Cineole	1031	1026	0.9	0.4
Benzyl alcohol	1037	1026	0.4	0.5
γ -Terpinene	1057	1054	0.1	0.1
<i>cis</i> -Linalool oxide (furanoid)	1072	1067	12.2	7.3
<i>trans</i> -Linalool oxide (furanoid)	1088	1084	7.0	7.5
Linalool	1098	1095	13.6	14.3
<i>trans</i> -Sabinene hydrate	1102	1098	t	0.1
Phenyl ethyl alcohol	1112	1107	4.5	6.5
<i>cis</i> -p-Menth-2-en-1-ol	1121	1118	1.0	1.2
<i>trans</i> -p-Menth-2-en-1-ol	1137	1136	0.5	0.5
<i>trans</i> -Verbenol	1148	1140	0.6	0.4
<i>cis</i> -Linalool oxide (pyranoid)	1168	1170	7.2	3.3
<i>trans</i> -Linalool oxide (pyranoid)	1172	1173	4.7	3.8
Terpinen-4-ol	1179	1174	7.9	10.6
<i>p</i> -Cymen-8-ol	1184	1179	0.7	0.5
α -Terpineol	1191	1186	6.3	6.6
Geraniol	1250	1249	0.9	1.3
<i>p</i> -Vinyl guaiaicol	1312	1309	2.0	0.7
Eugenol	1356	1356	0.1	0.1
1-Tetradecene	1388	1388	0.4	0.5
Geranyl acetone	1450	1453	0.8	1.2
2,4-Di-tert-butylphenol	1509	1512	1.5	1.6
1-Hexadecene	1589	1588	0.5	0.6
1-Octadecene	1789	1789	0.5	0.5
(<i>E,E</i>)-Farnesyl acetate	1838	1845	0.2	0.3
Hexahydrofarnesyl acetone	1843	1846	1.7	3.4
Methyl hexanoate	1923	1921	0.3	0.4
1-Eicosene	1990	1987	0.4	0.5
<i>n</i> -Heneicosane	2097	2100	0.9	0.8
(<i>E</i>)-Phytol	2112	2113	0.7	0.9
1-Docasene	2190	2189	0.3	0.4
<i>n</i> -Docasane	2196	2200	0.2	0.2
<i>n</i> -Tricosane	2296	2300	1.0	1.5
<i>n</i> -Tetracosane	2396	2400	0.2	0.2
<i>n</i> -Pentacosane	2496	2500	0.4	0.6
<i>Class composition</i>				
Monoterpene hydrocarbons			0.1	0.1
Oxygenated monoterpenes			66.7	60.3
Oxygenated sesquiterpenes			1.9	3.7
Oxygenated diterpenes			0.7	0.9
Benzenoids			9.2	9.9
Alkanes			2.7	3.3
Alkenes			2.1	2.5
Fatty acid ester			0.3	0.4
Total			83.7	81.1

^aRetention Index determined on BP-5 gas chromatography column (30 m x 0.25 mm) using *n*-alkanes; ^bretention index from literature; ^cmean of two samples; t: trace (<0.05%).

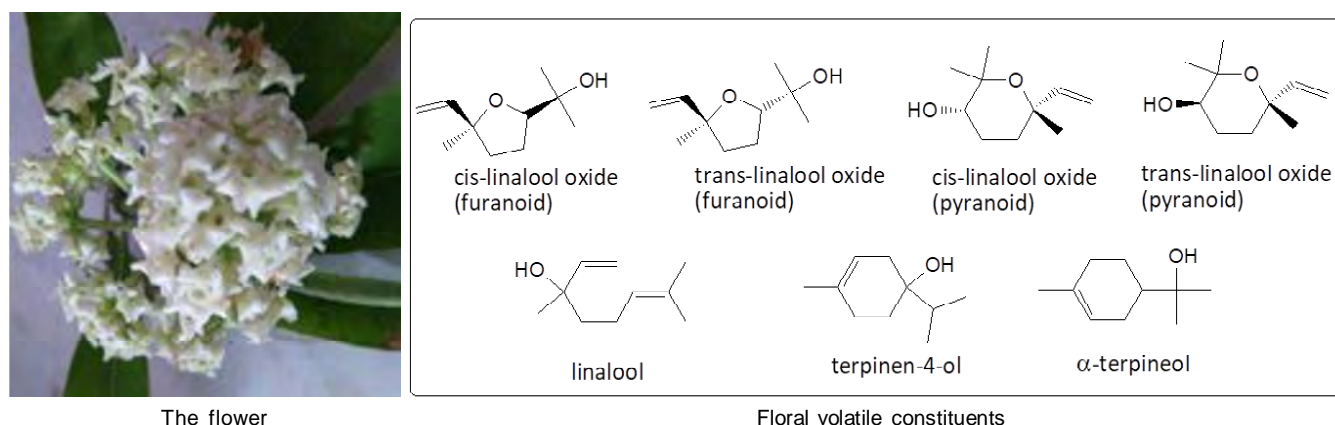


Figure 1: Major floral volatiles of *Alstonia scholaris*

the evening times in the aromatic plants (Toncer et al., 2016; Karik et al., 2019). It is reported that the decrease of essential oil content of flowers had a relationship with the increase in atmospheric temperature. In a study on *Rosa damascena* maximum content of the essential oil was obtained when the flowers picked up early morning (Kumar et al., 2013). Previously, the floral volatile oil composition of *A. scholaris* has been studied from Vietnam. The study reported linalool, *cis*- and *trans*-linalool oxides, α -terpineol, 2-phenylethyl acetate, and terpinen-4-ol as the major components of the volatile concentrate of *A. scholaris* (Dung et al., 2001). Besides, chemical resemblance, the amounts of *cis*- and *trans*-linalool oxides (31.1 % at 6 am and 21.9 % at 6 pm), terpinen-4-ol (7.9 % at 6 am and 10.6 % at 6 pm), and phenyl ethyl alcohol (4.5 % at 6 am and 6.5 % at 6 pm) was higher in the present study than Vietnam. However, the amount of linalool (13.6 % at 6 am and 14.3 % at 6 pm) and α -terpineol (6.3 % at 6 am and 6.6 % at 6 pm) was relatively lower in the present study than Vietnam. One of the major components in the Vietnam study, 2-phenyl ethyl acetate was not found in the present study. Despite the presence of several common components, the examined floral volatile oil of *A. scholaris* was quite different from those reported in the literature. These variations are might be a reflection of variation in various exogenous and endogenous factors (Barra, 2009). Further, this study identified four forms of linalool oxide (*cis*- and *trans*- forms of furanoid and pyranoid linalool oxide) in *A. scholaris* floral volatiles

that collectively made up to 31.1 % of total oil composition. The occurrence of furanoid linalool oxide is quite common in nature, while the prevalence of pyranoid linalool oxide is less (Kumar et al., 2013). Furanoid linalool oxide plays a major role in the smell of linalool oxide, which has leafy, earthy, sweet, floral, and creamy odour. Pyranoid linalool oxide has a sweet, floral, creamy, and earthy odour. They are of high commercial importance in perfumery (Wang et al., 1994; Qin and Yang, 2016). The linalool is a much sought-after compound in the flavour and fragrance industry. This compound is also present in the floral fragrance of many plant families such as Lamiaceae, Lauraceae, Verbenaceae, etc. and is attractive to a broad spectrum of pollinators, herbivores, and parasitoids (Kamatou and Viljoen, 2008).

The floral volatile oils of *A. scholaris* were screened for antibacterial activity against Gram-positive and Gram-negative bacterial strains. The Gram-negative bacterial strains, namely *S. typhimurium* (MTCC-98) and *E. coli* (MTCC-739) were found to be susceptible to the essential oils (ZOI: 10-11 mm and 7 mm, respectively). However, the oils exhibited low antibacterial activity against Gram positive bacterial species tested (ZOI: nil-2.0 mm). In general, morning and evening collected floral volatile oils exhibited comparable antibacterial activity. Slight variation in the activity may be due to compositional variation in these oils. The observed antibacterial activity might be due to the presence of a higher amount of oxygenated monoterpenes (60.3-66.7 %) in the oil. Other major or minor

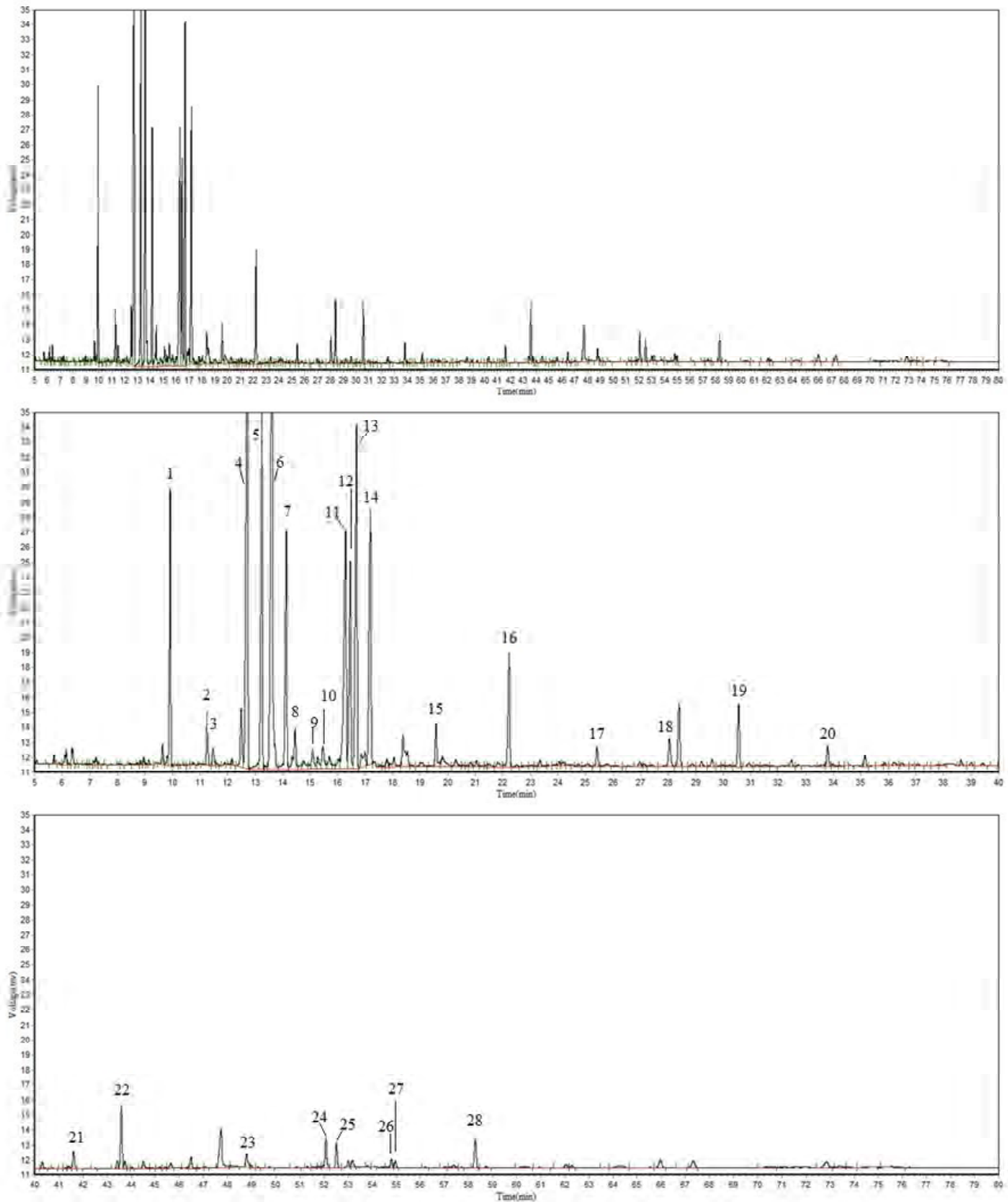


Figure 2: Gas chromatogram (GC-FID) of the floral volatile oil of *Alstonia scholaris*. Peak 1: dehydro-1,8-cineole; 2: 1,8-cineole; 3: benzyl alcohol; 4: *cis*-linalool oxide (furanoid); 5: *trans*-linalool oxide (furanoid); 6: linalool; 7: phenyl ethyl alcohol; 8: *cis*-p-menth-2-en-1-ol; 9: *trans*-p-menth-2-en-1-ol; 10: *trans*-verbenol; 11: *cis*-linalool oxide (pyranoid); 12: *trans*-linalool oxide (pyranoid); 13: terpinen-4-ol; 14: α -terpineol; 15: geraniol; 16: *p*-vinyl guaiacol; 17: 1-tetradecene; 18: geranyl acetone; 19: 2,4-di-tert-butylphenol; 20: 1-hexadecene; 21: 1-octadecene; 22: hexahydrofarnesyl acetone; 23: 1-eicosene; 24: *n*-heneicosane; 25: (*E*)-phytol; 26: 1-docasene; 27: *n*-docasane; 28: *n*-tricosane.

constituents present in the oils might be also responsible for the observed antimicrobial activity (Kamatou and Viljoen, 2008; Guimaraes et al., 2019).

CONCLUSION

It can be concluded that the collection time influenced the yield and chemical composition of *A. scholaris* floral volatiles. Evening (6 pm) collection of flowers can be beneficial for getting a higher yield of floral volatile oil. However, morning (6 am) collection can be beneficial for getting a linalool oxides rich floral volatile oil.

ACKNOWLEDGEMENTS

Authors are thankful to the CSIR, New Delhi for financial support (Project: HCP-0007) and to Dr. Amit Chauhan (Plant Taxonomist), CSIR-CIMAP, Research Centre Pantnagar for identification of the plant material. This is a CIMAP Communication No.: CIMAP/PUB/2020/OCT/107

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